

## Photolysis of the Nitrite Esters of 4,4-Dimethyl-6 $\beta$ -hydroxy-steroids

By M. P. KULLBERG and B. GREEN\*

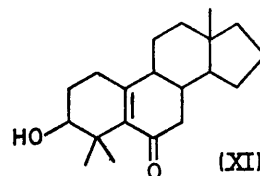
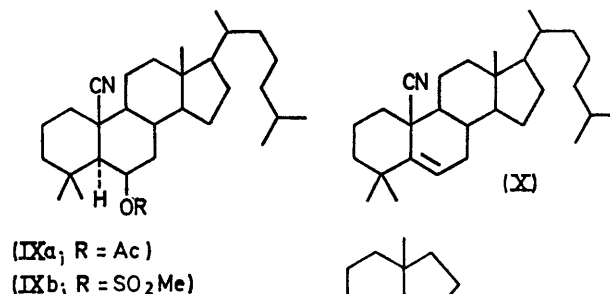
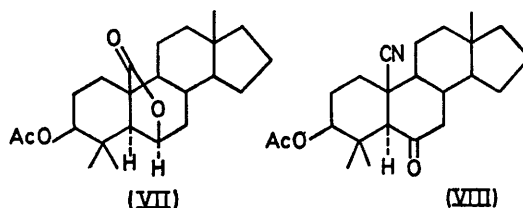
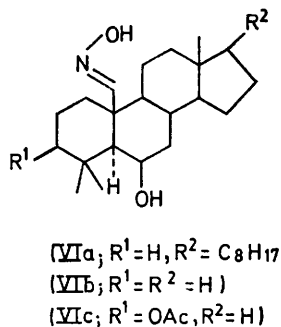
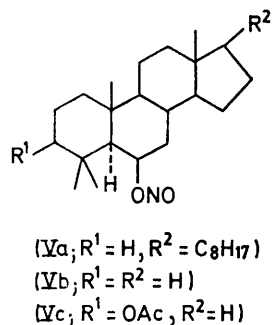
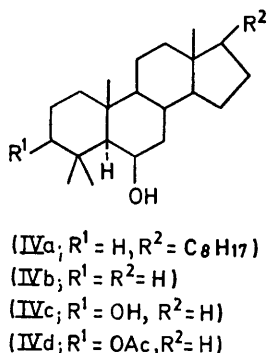
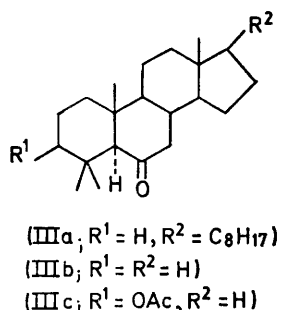
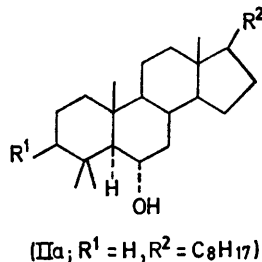
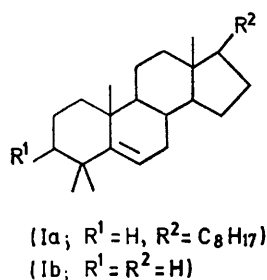
(Chemistry Department, University of Maine, Orono, Maine)

**Summary** Photolysis of the nitrite esters of three 4,4-dimethyl-6 $\beta$ -hydroxy-steroids leads to exclusive functionalization at C-19; the stereochemical implications are discussed.

ALTHOUGH considerable work has been carried out on the effects of triaxial interactions in the steroid A-ring<sup>1</sup> almost no study has been devoted to the situation involving one of the groups in the B-ring.† We have investigated the abstraction by 6 $\beta$ -alkoxy radicals of hydrogen atoms from the 4 $\beta$  (31) and 10 $\beta$  (19) methyl groups of 4,4-dimethyl steroids.<sup>2,3</sup>

m.p. 108–109°,  $\nu_{\max}$  1710 cm<sup>-1</sup>. Reduction with (LiAlH<sub>4</sub>; 24 h; 35°) gave in high yield (IVa), m.p. 106–107°;  $\delta$  4.30 p.p.m. (6 $\alpha$ -H,  $W_{\frac{1}{2}}$  6 Hz). Analogously (Ib) was converted into (IVb), m.p. 125–126°,  $\delta$  4.40 p.p.m. (6 $\alpha$ -H,  $W_{\frac{1}{2}}$  7 Hz). (IVd), m.p. 134–135°,  $\nu_{\max}$  3600 and 1740 cm<sup>-1</sup>,  $\delta$  4.40 p.p.m. (3 $\alpha$ -H, 6 $\alpha$ -H) was obtained by selective acetylation of the 3 $\beta$ ,6 $\beta$ -diol (IVc), m.p. 157–159°, prepared by prolonged reduction (LiAlH<sub>4</sub>) of (IIIc).<sup>4</sup> Alternatively (IVd) was prepared directly from 4,4-dimethyl-androst-5-en-3-one.<sup>4</sup>

Photolyses of the nitrite esters (Va and b) from alcohols (IVa and b), proceeded similarly to furnish the 19-oximes (VIa and b), m.p. 211–212 and 209–210°, in yields of 44 and 42% respectively, together with small amounts of the 6 $\beta$ -alcohols (IVa and b), and the 6-ketones (IIIa and b). No trace of the 31-oximes could be found.



Photolysis of the nitrite ester (Vc) gave a more complex mixture, from which the 6 $\beta$ -alcohol (IVd) and 6-ketone (IIIc) were separated in small amount from the 19-oxime (VIc) (42%), m.p. 207–209°. The most polar chromatography fraction consisted of a mixture of 19-oxime (VIc) and the 6 $\beta$ ,19-lactone (VII), m.p. 232–234°, whose identity was confirmed by its preparation from oxime (VIc) by chromic oxide-acetone oxidation.<sup>5</sup> Repeated photolyses revealed no trace of the 31-oxime.

Further transformations were carried out to confirm the 19- as opposed to 31-functionalisation. (VIc) was oxidized (CrO<sub>3</sub>-pyridine) to (VIII), m.p. 250–251°. In refluxing dilute methanolic KOH (VIII) was converted into the

Prolonged hydroboration of (Ia), followed by Jones oxidation of the resulting 5 $\alpha$ -H,6 $\alpha$ -ol (IIa) gave (IIIa),

† Whalley and his co-workers are engaged in similar complementary studies.

$\alpha\beta$ -unsaturated ketone (XI). (VIa) was converted by sodium acetate in refluxing acetic anhydride into (IXa), m.p. 137—138°. Finally, (VIa) was converted *via* the cyano-mesylate (IXb) into (X), m.p. 88—90°.

By analysis of the chemical shift values of the 18- and 31-methyl groups in the cyano-compounds [(VIII):  $\delta$  0.82 (Me-18), 1.50 (Me-31); (IXa): 0.79 (Me-18), 1.19 (Me-31); (X) 0.78 (Me-18), 1.32 (Me-31) p.p.m.] and consideration of the spatial shielding characteristics of the CN group,<sup>6,7</sup> the 10 $\beta$ -location of the CN group has been proved conclusively.

The abstraction by a 6 $\beta$ -alkoxy-radical of a hydrogen

atom exclusively from the 19-methyl group in compounds (IVa, b, and d) indicates, as in Whalley's studies, that A-ring conformations other than ideal chairs are involved. In the case of the 3 $\beta$ -acetoxy steroid (IVd) a flattened chair is indicated, since a boat or twist conformation would be ruled out by 3 $\beta$ -acetoxy-19-methyl interaction. In the 3-deoxy series either a flattened chair or twist conformation is consistent with the results.

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<sup>1</sup> D. L. Robinson and D. W. Theobald, *Quart. Rev.*, 1967, **21**, 314.

<sup>2</sup> D. H. R. Barton, J. M. Beaton, L. E. Geller, and M. M. Pechet, *J. Amer. Chem. Soc.*, 1960, **82**, 2640; D. H. R. Barton and J. M. Beaton, *ibid.*, 1960, **82**, 2641.

<sup>3</sup> J. M. Midgley, J. E. Parkin, and W. B. Whalley, *Chem. Comm.*, 1970, 789.

<sup>4</sup> C. R. Eck, M. P. Kullberg, and B. Green, *Chem. Comm.*, 1971, in the press.

<sup>5</sup> D. H. R. Barton and J. M. Beaton, *J. Amer. Chem. Soc.*, 1962, **84**, 199.

<sup>6</sup> G. S. Reddy, J. M. Goldstein, and L. Mandell, *J. Amer. Chem. Soc.*, 1961, **83**, 1300.

<sup>7</sup> A. D. Cross and I. T. Harrison, *J. Amer. Chem. Soc.*, 1963, **85**, 3223.